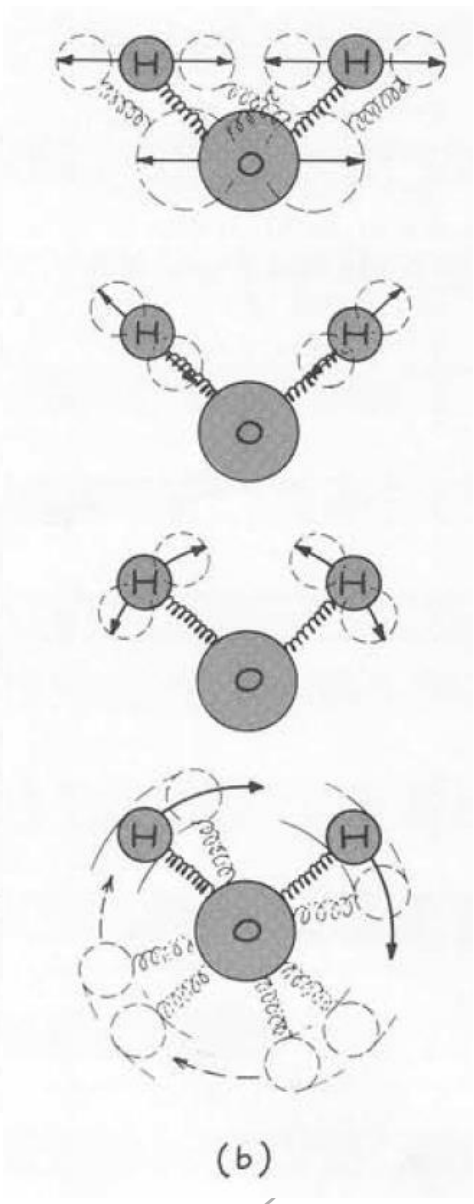
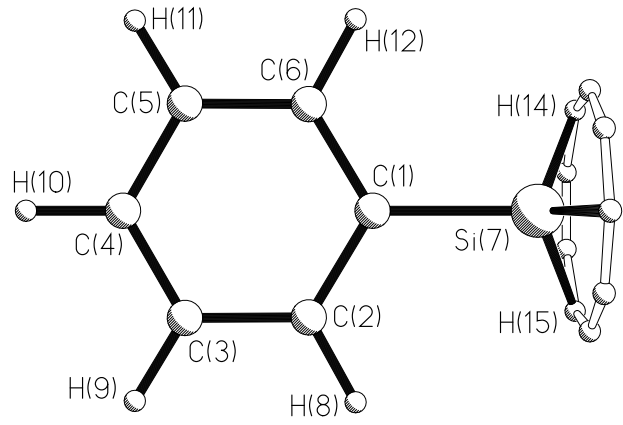
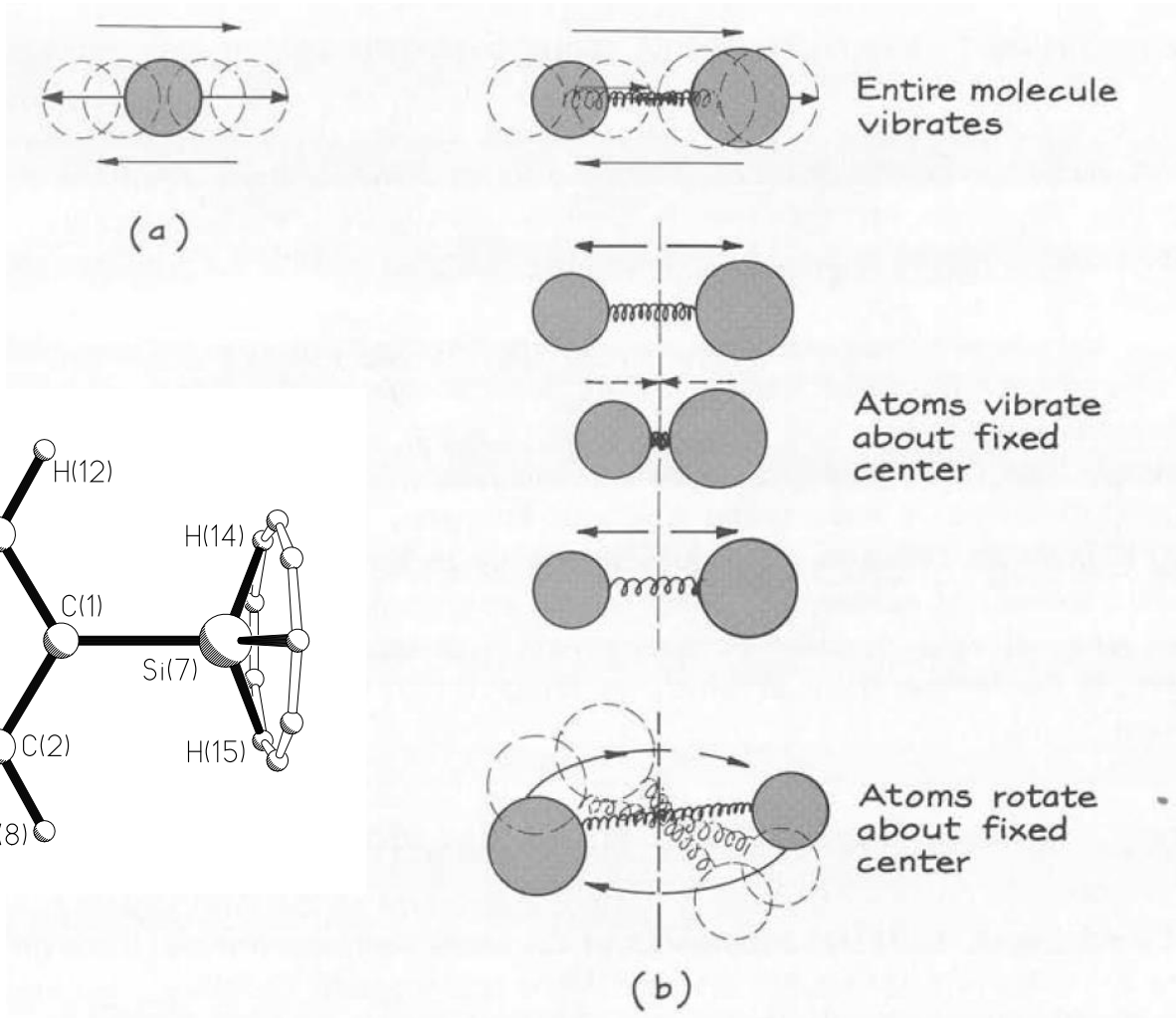


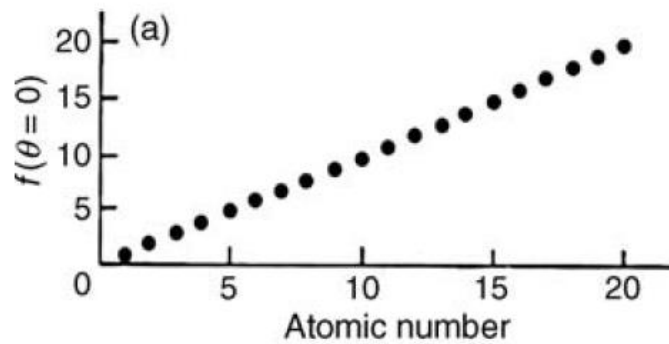
Structural Methods

Information	Gas	Liquid	Solid
Fingerprint	IR, (<i>MW</i> *)	IR, Raman	IR, Raman
	MS, UV/Vis	MS, UV/Vis	MS, UV/Vis
	NMR		Pulver-XRD
Groups	IR, MS, NMR	IR, Raman	IR, Raman
		MS, NMR	MS, *S-NMR, (* <i>Mössbr.</i>)
Molecular	IR, (* <i>MW</i>)	IR, Raman	IR, Raman, S-NMR,
Symmetry	**GED	NMR	SC-XRD, (**NQR)
Lengths,	* <i>MW</i> , **GED	EXAFS***	SC-XRD,
Angles	IR, Raman	(<i>NMR</i>), (<i>XRD</i>)	***ND
Electronic	UV/Vis	UV/Vis	UV/Vis, PES, XPS, XRD
Structure	PES, ESR	ESR, (*XPS)	* <i>Mössbr.</i> , ***ND
Electron density			SC-XRD

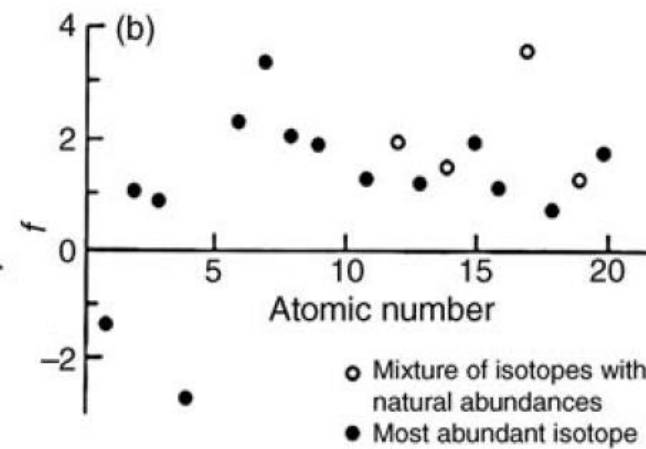
Vibrational Average in Single Crystal Determination



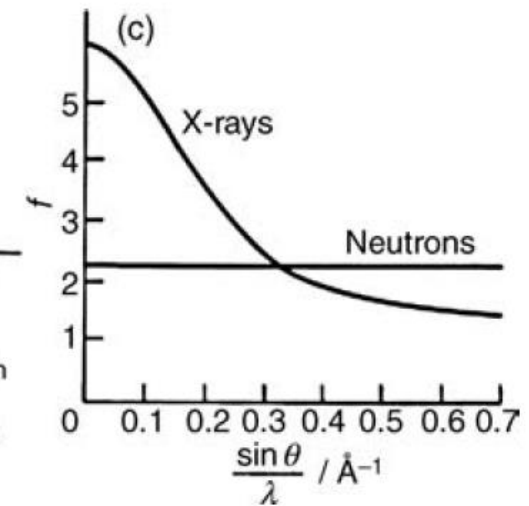
Variation of the scattering power scattering amplitude f with the atomic number



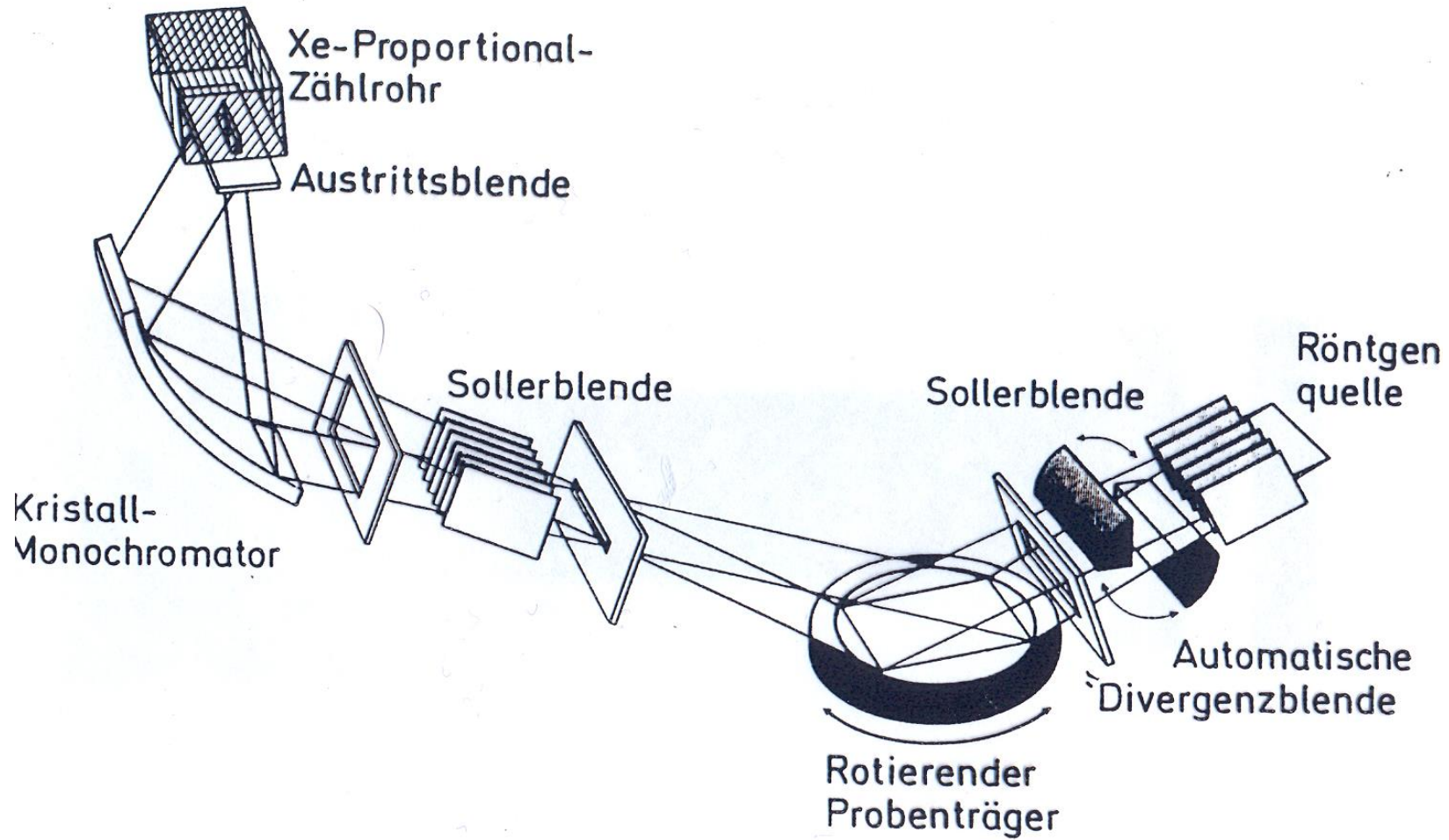
X-rays

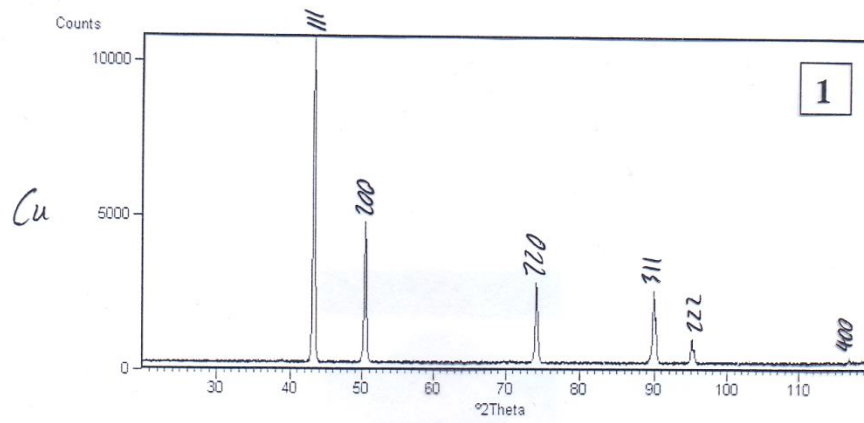


neutrons

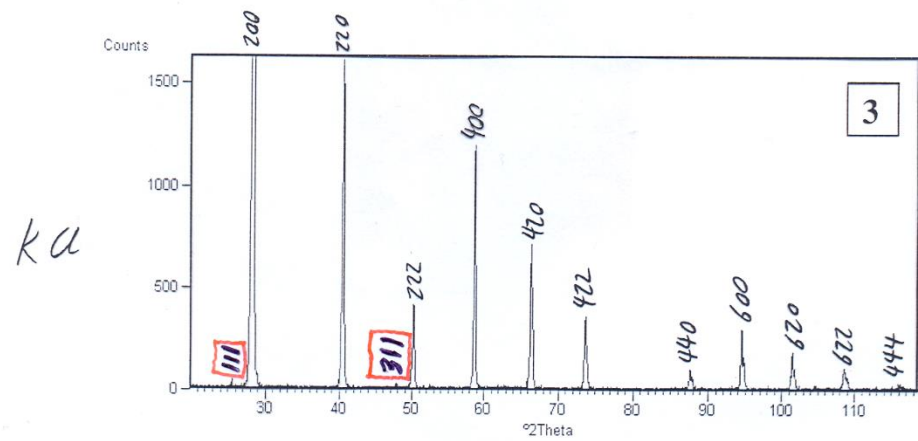
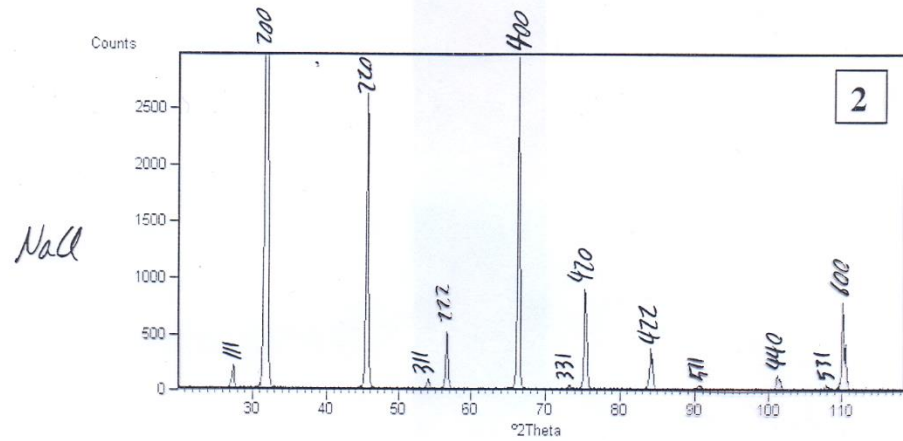


Pulverdiffraktometer





Diffraktogramme von
Cu, NaCl und KCl



Diffraktogramme von Phasengemischen

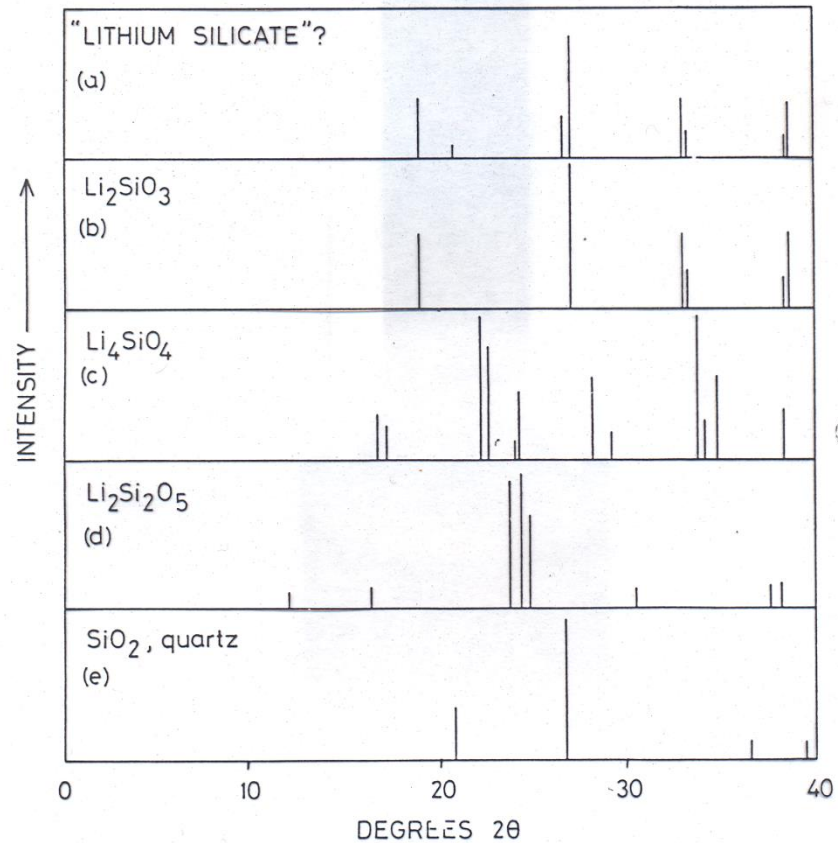


Fig. 5.47 Powder patterns of (a) a bottle labelled 'lithium silicate' and (b) to (e) standard lithium silicate and silica phases

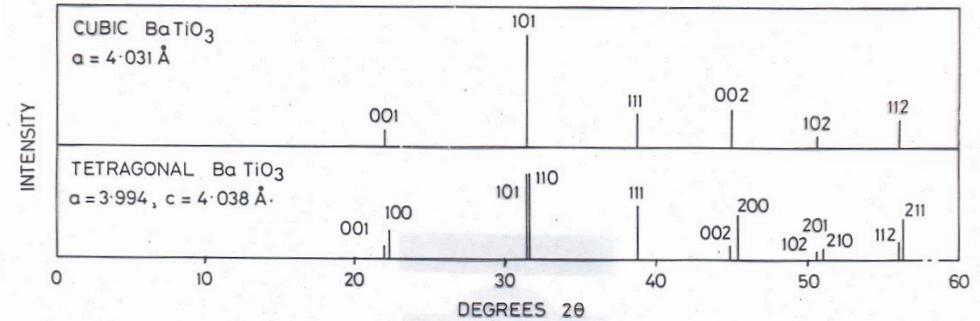
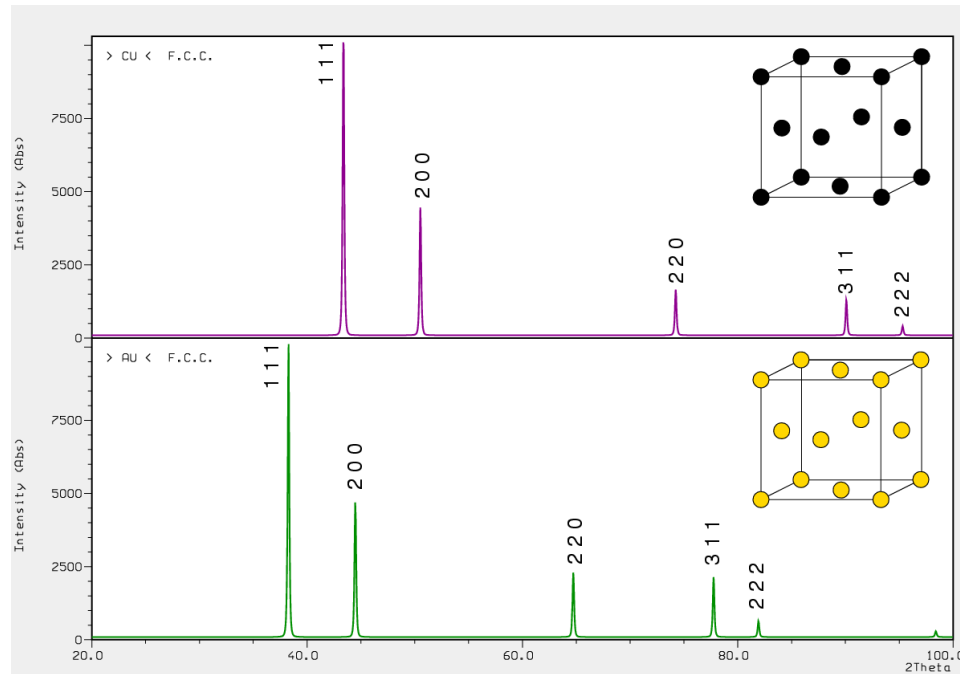


Fig. 5.46 Powder patterns of cubic and tetragonal BaTiO_3 , showing the influence of crystal symmetry and multiplicities on the number of lines that are observed.

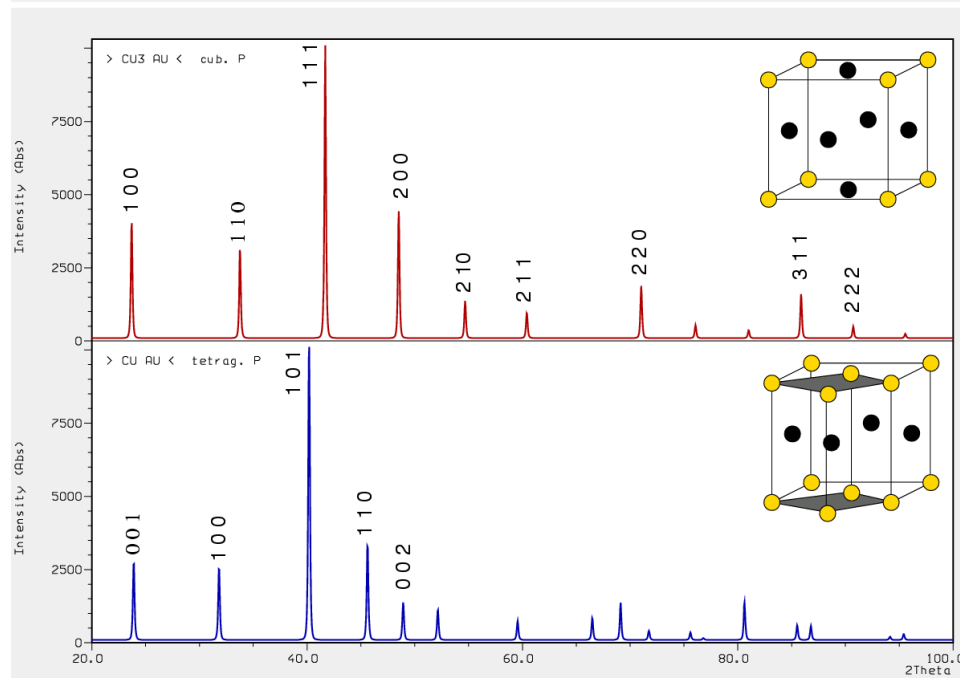
Pulverdiagramme des Systems Cu/Au

ruby.chemie.uni-freiburg.de/Vorlesung/methoden_II_4.xhtml



- Kupfer (fcc / kdp)

- Gold (fcc / kdp)



- Cu₃Au (Kub. P)

- CuAu (tetr. P)

Monoklin

$$\sin^2 \theta = \frac{\lambda^2}{4} \left[\frac{h^2}{a^2 \sin^2 \beta} + \frac{k^2}{b^2} + \frac{l^2}{c^2 \sin^2 \beta} - \frac{2hl \cos \beta}{ac \sin^2 \beta} \right]$$

Orthorhombisch

$$\sin^2 \theta = \frac{\lambda^2}{4} \left[\frac{h^2}{a^2} + \frac{k^2}{b^2} + \frac{l^2}{c^2} \right]$$

Tetragonal

$$\sin^2 \theta = \frac{\lambda^2}{4a^2} \left[h^2 + k^2 + \left(\frac{a}{c}\right)^2 l^2 \right]$$

Hexagonal und trigonal

$$\sin^2 \theta = \frac{\lambda^2}{4a^2} \left[\frac{4}{3} (h^2 + k^2 + hk) + \left(\frac{a}{c}\right)^2 l^2 \right]$$

Kubisch

$$\sin^2 \theta = \frac{\lambda^2}{4a^2} \left[h^2 + k^2 + l^2 \right]$$

Triklin

$$\sin^2 \theta = \frac{\lambda^2}{4} [h^2 a^{*2} + k^2 b^{*2} + l^2 c^{*2} + 2klb^*c^* \cos \alpha^* + 2lhc^*a^* \cos \beta^* + 2hka^*b^* \cos \gamma^*]$$

$$a^* = \frac{1}{V} bc \sin \alpha, \quad \cos \alpha^* = \frac{\cos \beta \cos \gamma - \cos \alpha}{\sin \beta \sin \gamma}$$

$$b^* = \frac{1}{V} ca \sin \beta, \quad \cos \beta^* = \frac{\cos \gamma \cos \alpha - \cos \beta}{\sin \gamma \sin \alpha}$$

$$c^* = \frac{1}{V} ab \sin \gamma, \quad \cos \gamma^* = \frac{\cos \alpha \cos \beta - \cos \gamma}{\sin \alpha \sin \beta}$$

$$V = abc \sqrt{1 + 2 \cos \alpha \cos \beta \cos \gamma - \cos^2 \alpha - \cos^2 \beta - \cos^2 \gamma}$$

$$\Delta = \lambda/10$$

