

# Modeling Functionalities of LMPS Mozzarella based on Calcium Balances

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# Importance of Calcium in Cheese Functionality

- Total calcium is well known to influence cheese texture
- The insoluble calcium phosphate content is an important indicator of meltability (Lucey and Fox, 1993)
  - Previously thought that all calcium was solubilized at cheese pH (like milk)
- The insoluble calcium phosphate content of cheese decreases during ripening (Hassan et al., 2004)
  - Previously thought that calcium solubilization was done by end of manufacture (no further acid)
- Cheesemakers are often recommended to have more “early” acid if they want to effectively remove calcium to improve melt/curdiness.
  - **Why?**
- **How much calcium is lost at key points in the cheesemaking process?**
- **What are the kinetics of calcium phosphate solubilization during cheesemaking? Does it matter how fast the acid is produced?**



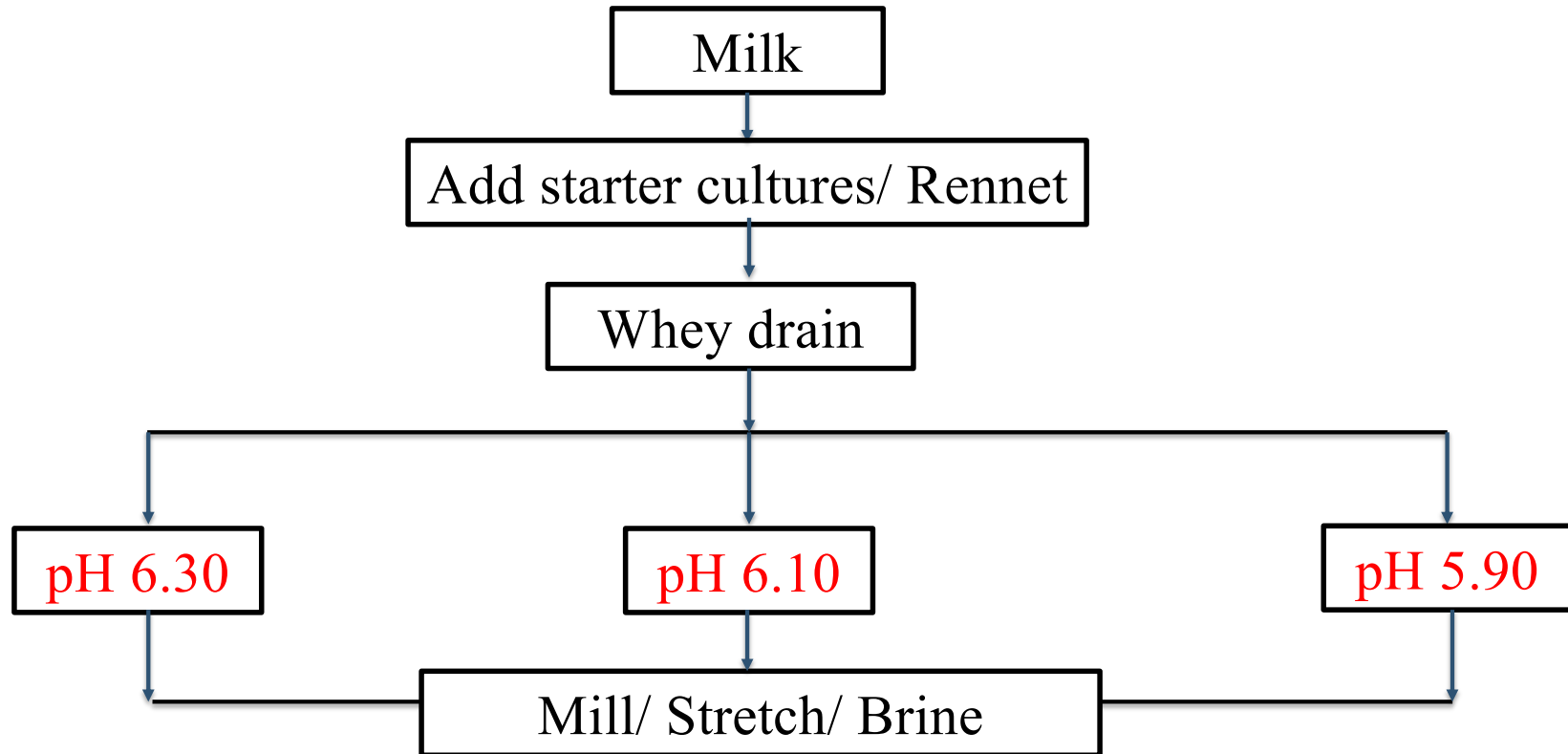
# Preliminary Cheese Make in Pilot plant

## Whey Drain Study



*CDR Pilot Plant*

# Experimental Design

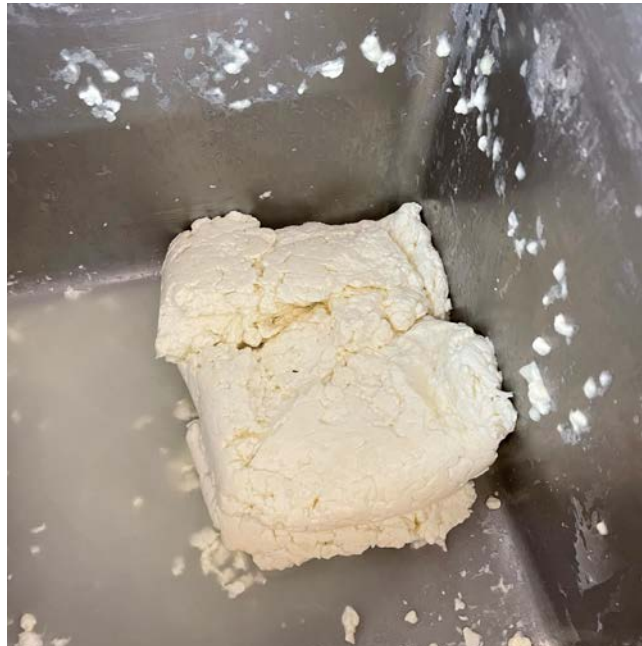


# LMPS Mozzarella Manufacture

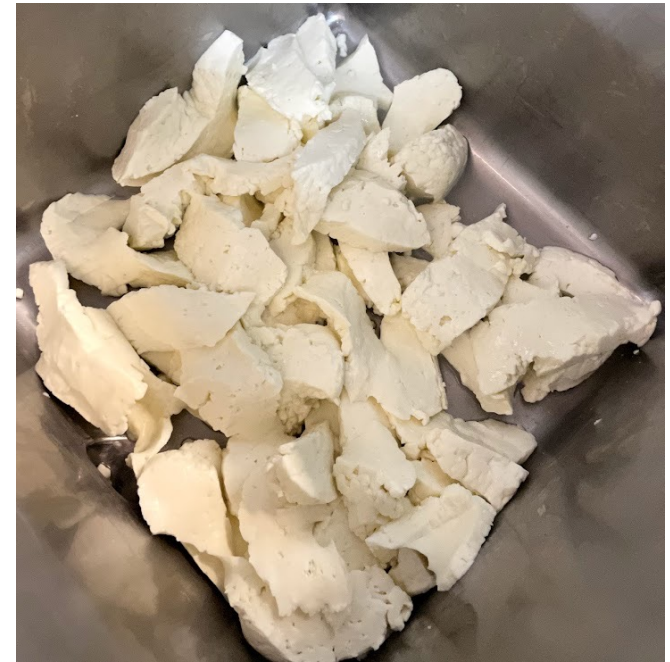
**1. Initial milk**



**2. After whey drain**



**3. Milled curd**

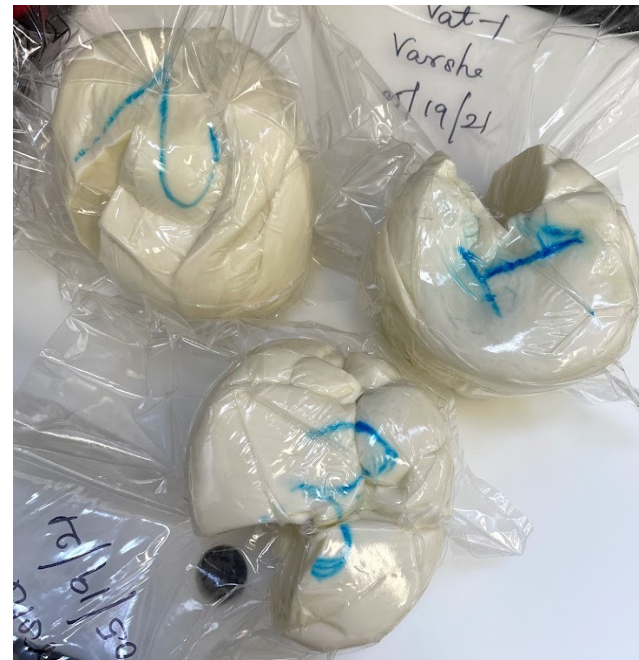


# LMPS Mozzarella Manufacture

## 4. Stretched curd



## 5. Curds after brining



# Analysis Area in the Pilot Plant

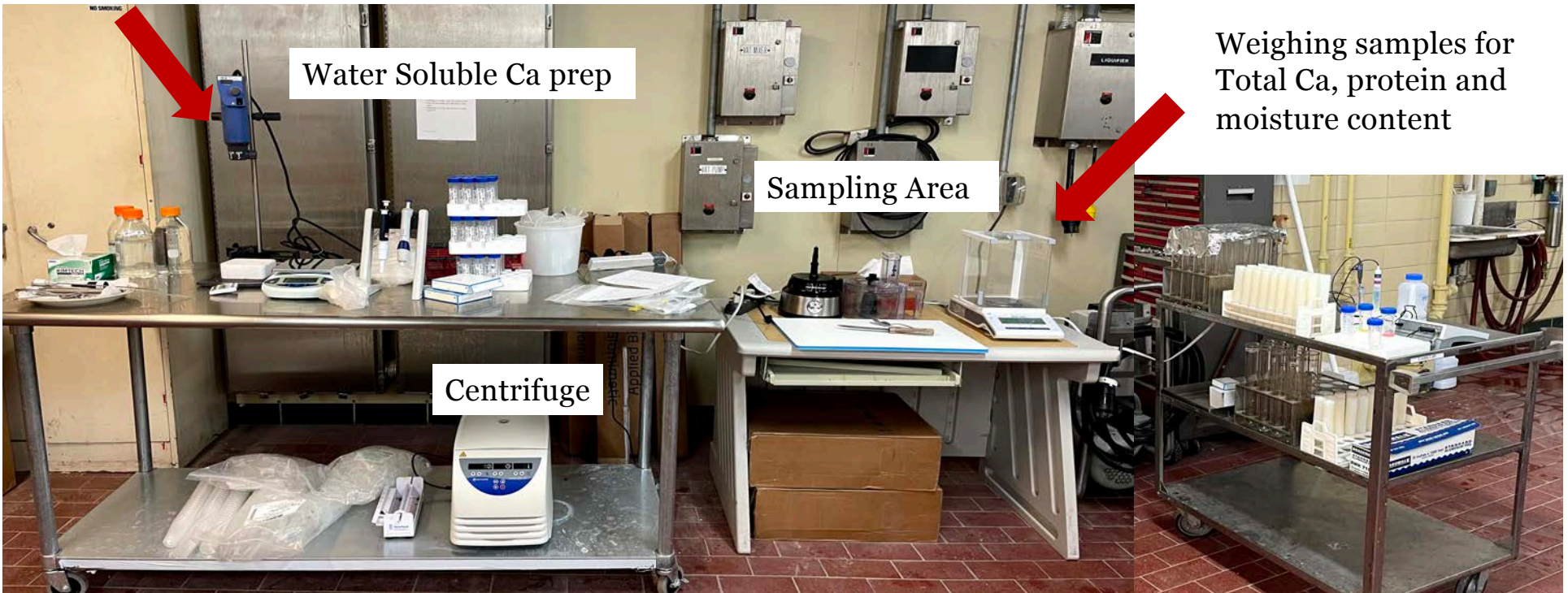
Hand-held mixer

Water Soluble Ca prep

Sampling Area

Weighing samples for Total Ca, protein and moisture content

Centrifuge



# Analysis Area





# Analyses

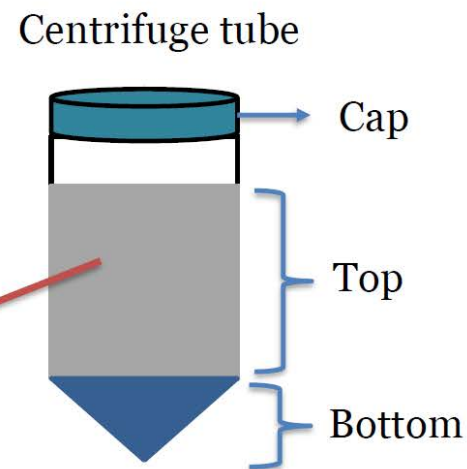
- Milk composition
- Cheese composition
- Titrations and soluble Ca in milk (rennet gel)
- Total and insoluble Ca in curds during manufacture-ICP
- Protein content-Kjeldahl
- Total solids-Loss on drying



# Method Developed to Monitor State of Ca during Cheesemaking

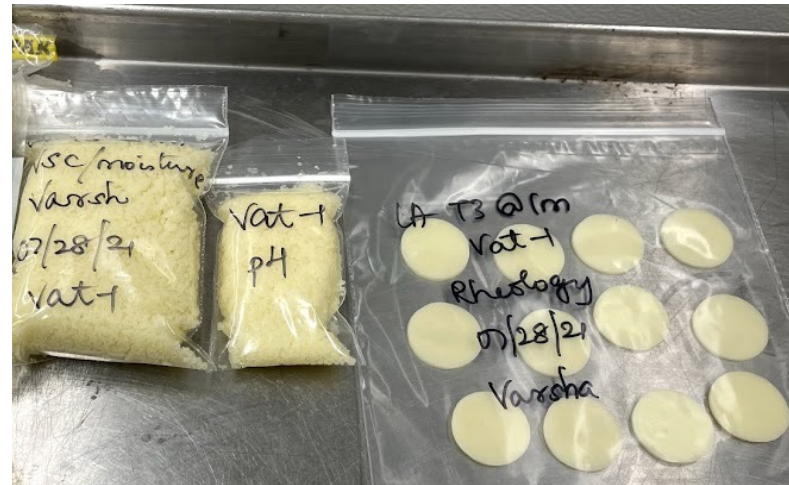
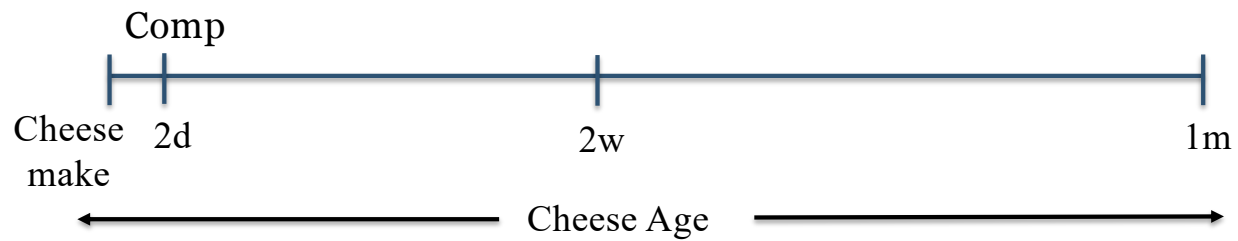
## Sample preparation

- 4 g of sampled curds in 40 g of water at temp-55°C
- Homogenized for 2 minutes
- Centrifuged for 10 minutes at 10,000 rpm
- Supernatant- Ca measured using ICP– **Water soluble calcium**

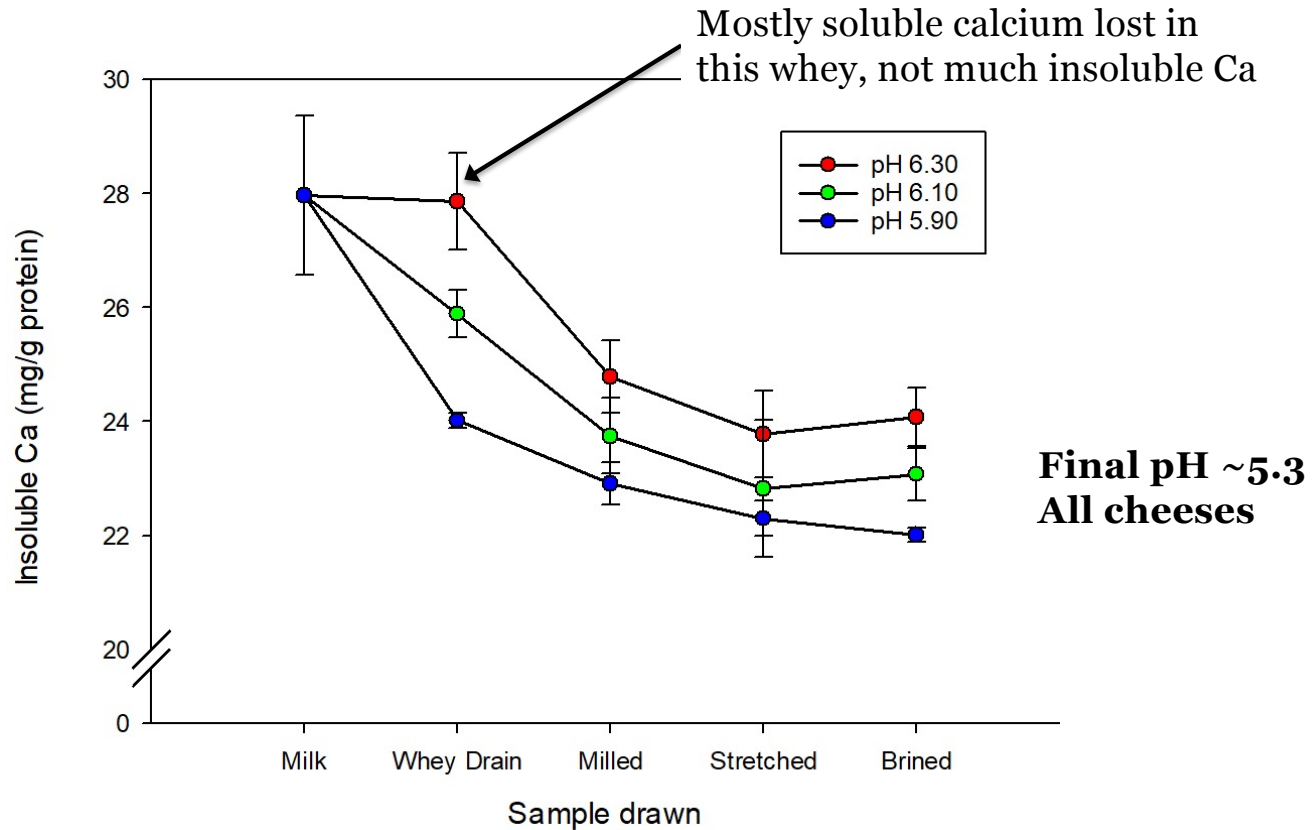


- Top layer- soluble Ca
- Bottom layer- Insoluble Ca

# Sampling and Analyses



# Calcium Balances During Manufacture



# Current Work

- Bench top experiments
  - To study the effects of rate of acidification on calcium balances in milk
  - Glucono-delta-lactone (GDL) would be used to acidify the milk
    - Same concentration of GDL would hydrolyze faster at higher temperatures (~40 °C) and acidify the milk faster as compared to lower temperatures
    - This property of GDL can be used to vary the rates of acidification in milk and to study its effects on calcium balances when pre-acidified to desired pH
  - Low heat non-fat dry milk (NFDm) as sample



# Initial trial- Rate of Acidification

10% NFD (Protein- 3.47%)

- GDL concentration- 0.05 %
- Temp (20, 30, 40°C)
- pH drop measured
- Samples were drawn at pH 6.30 and 6.10 at each of the temperatures for insoluble calcium measurement
  - Amicon 10 kDa centrifugal filters were used to extract the soluble calcium



# Need a rapid method to measure changes in soluble Ca during acidification of milk

- Amicon tubes with 10 kDa filter: quickly generates permeate

10000 × g for 5 min



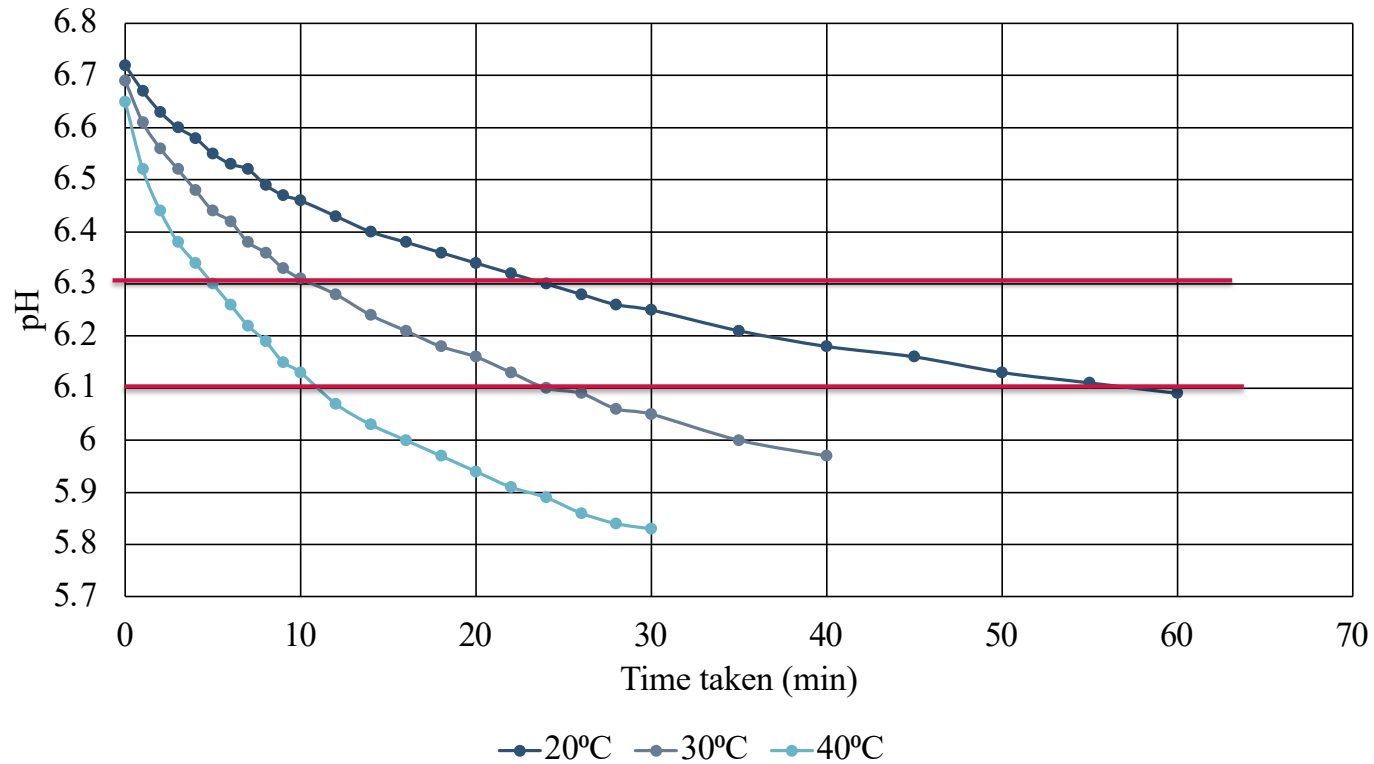
10000 × g for 10 min



15000 × g for 10 min



# Temperature vs Rate of acidification



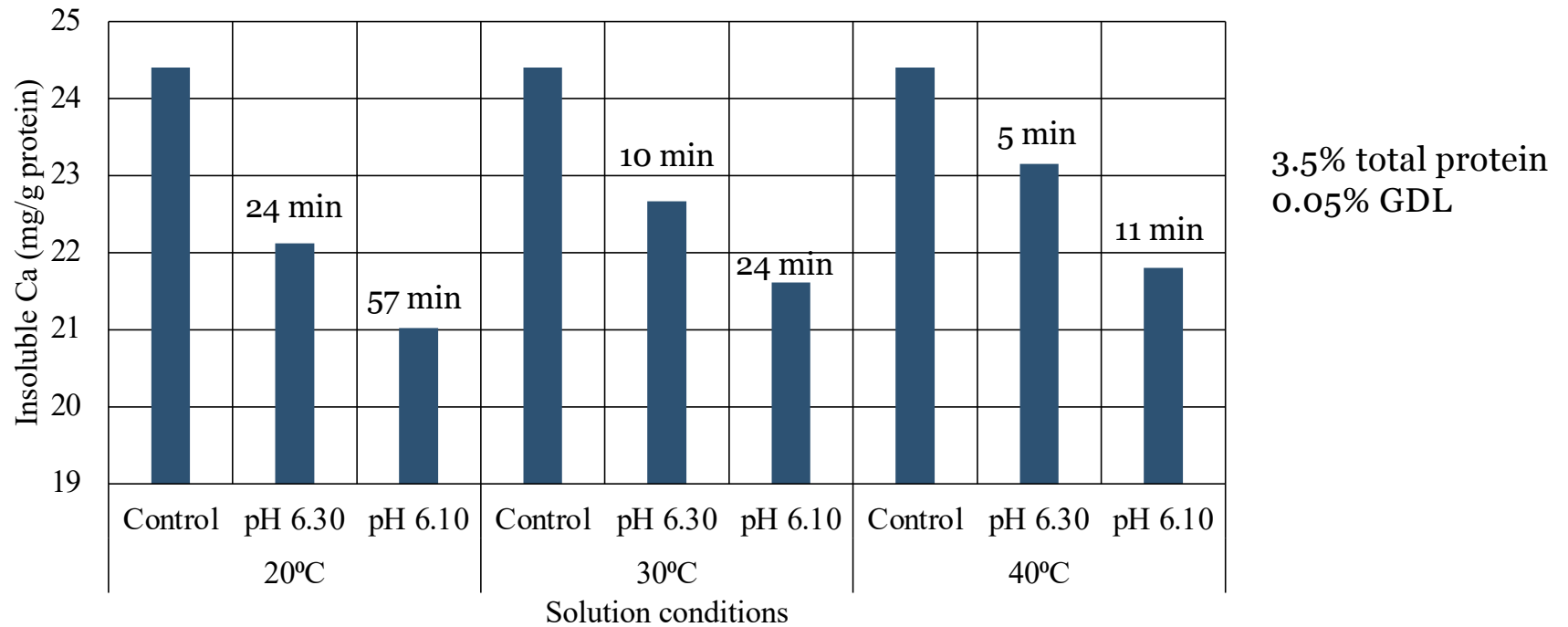
Samples were drawn when pH reached 6.30 and 6.10, respectively at each temperatures and the insoluble calcium was measured



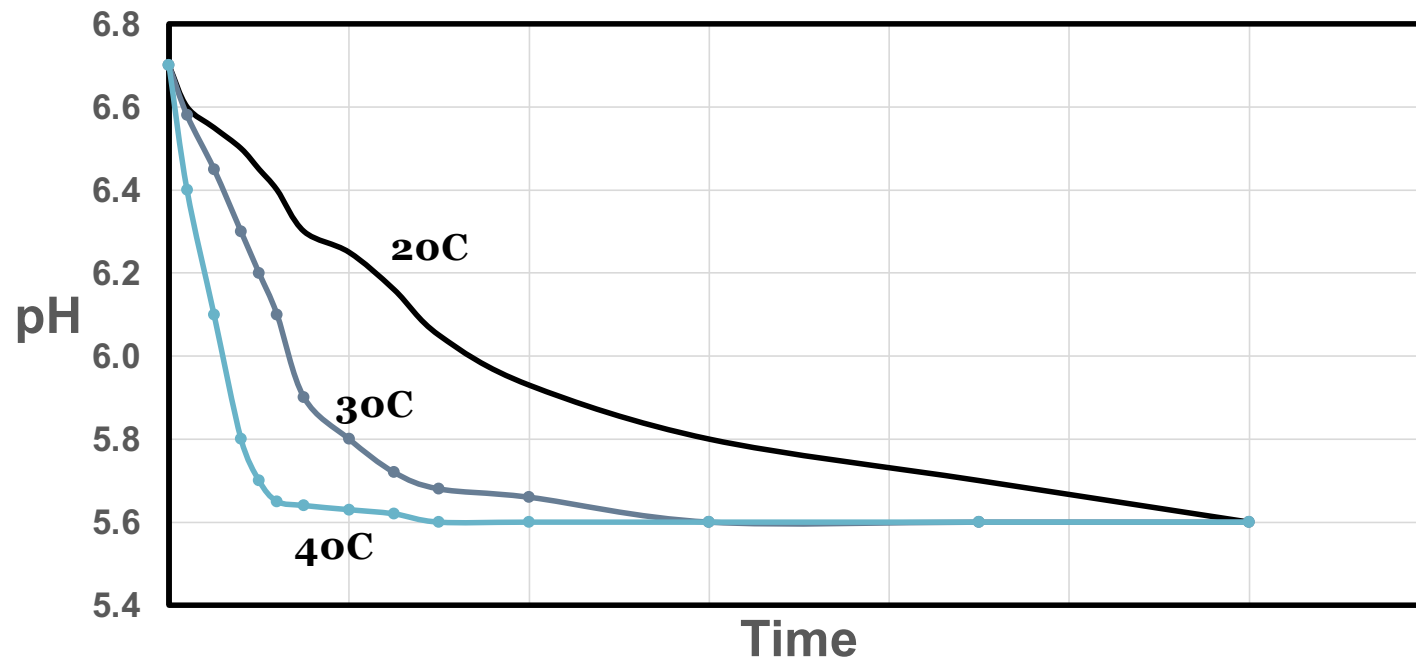


# Looked at Impact of Rate of Acidification on Insoluble Ca in Milk

## Slower Rate of Acidification Resulted in Greater Ca Solubilization



# Altering temperature, but using same GDL concentration, allows us to vary time to reach the same target pH



# Future Experiments

Study the effects of rate of acidification on calcium balances

- Constant GDL conc. (chosen from previous experiment)
- Various temperatures (20, 30, 40°C)
- Measure the insoluble calcium once the rate of acidification has reached a plateau (similar pH values reached at different temp since constant GDL conc.)
- Is there a difference in calcium balances with different rates of acidification?
- How does that impact cheese functionality?



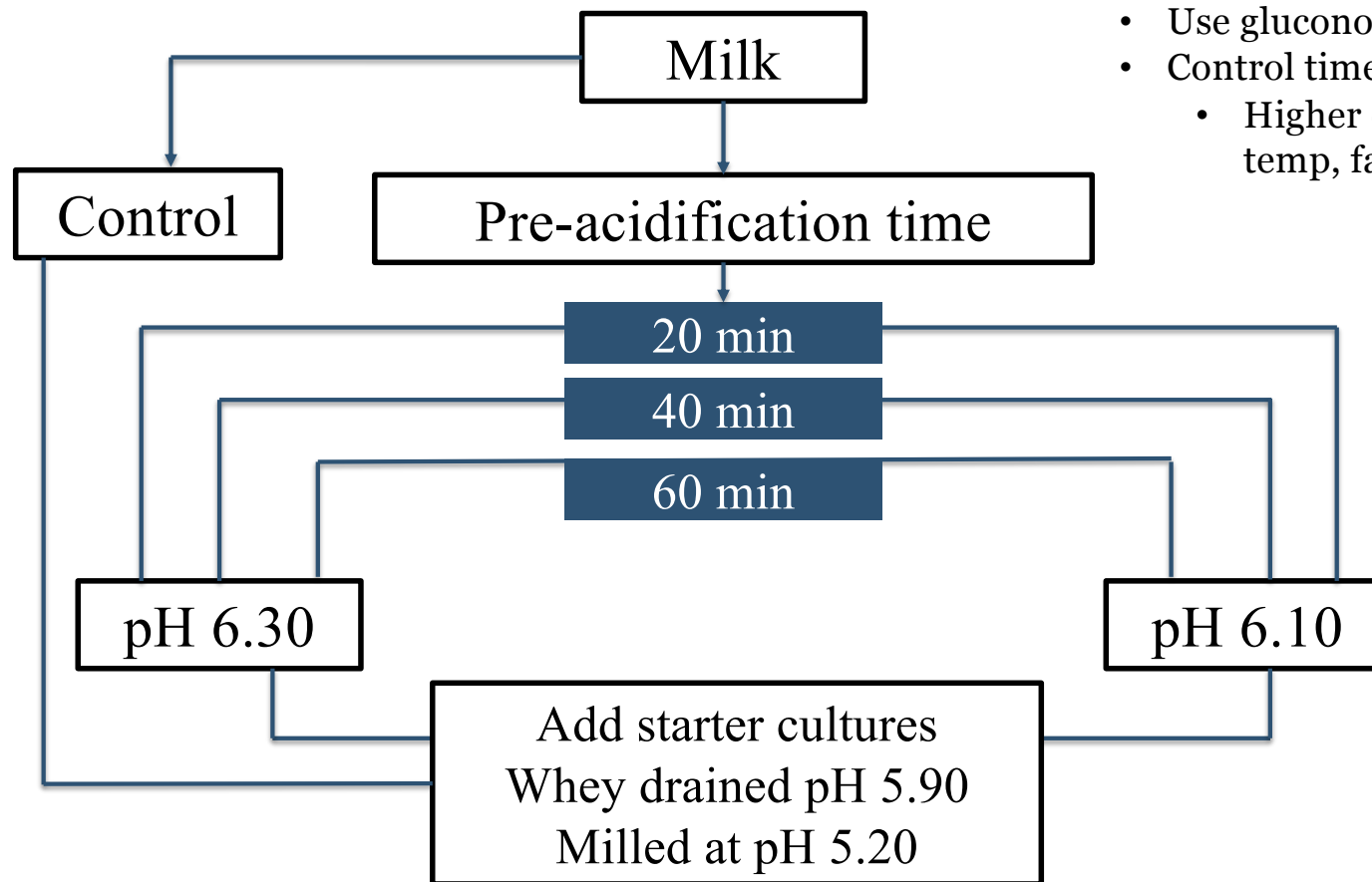
# Dairy plant Cheesemake Experiments

## Cheese making in dairy plant

- Experimental design
  - Rate of acidification vs pH
  - Different acids vs pre-acidification pH
  - Different casein conc. vs acids/ pre-acidification pH

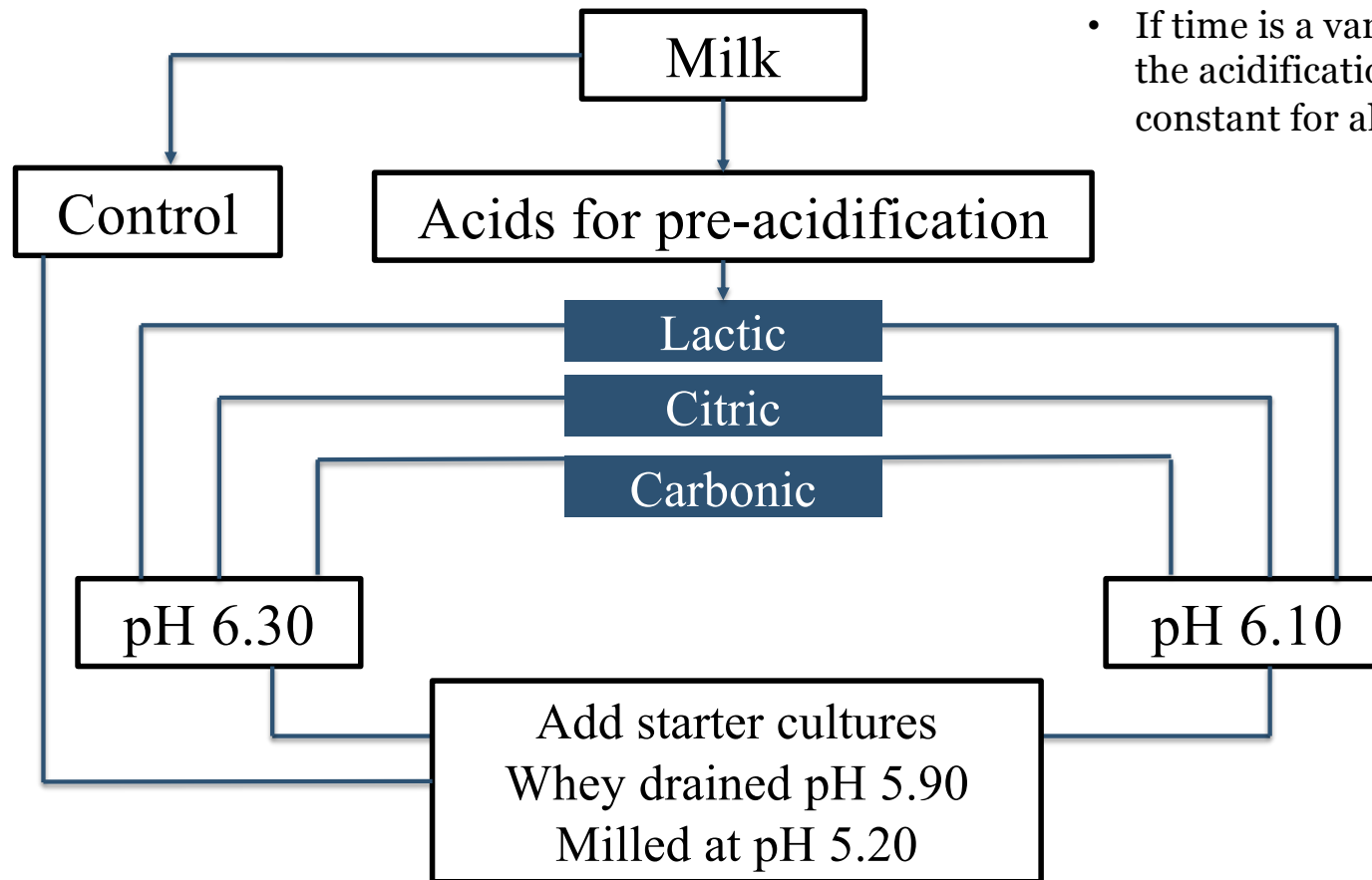


# Exp 1: Rate of acidification vs pre-acidification pH



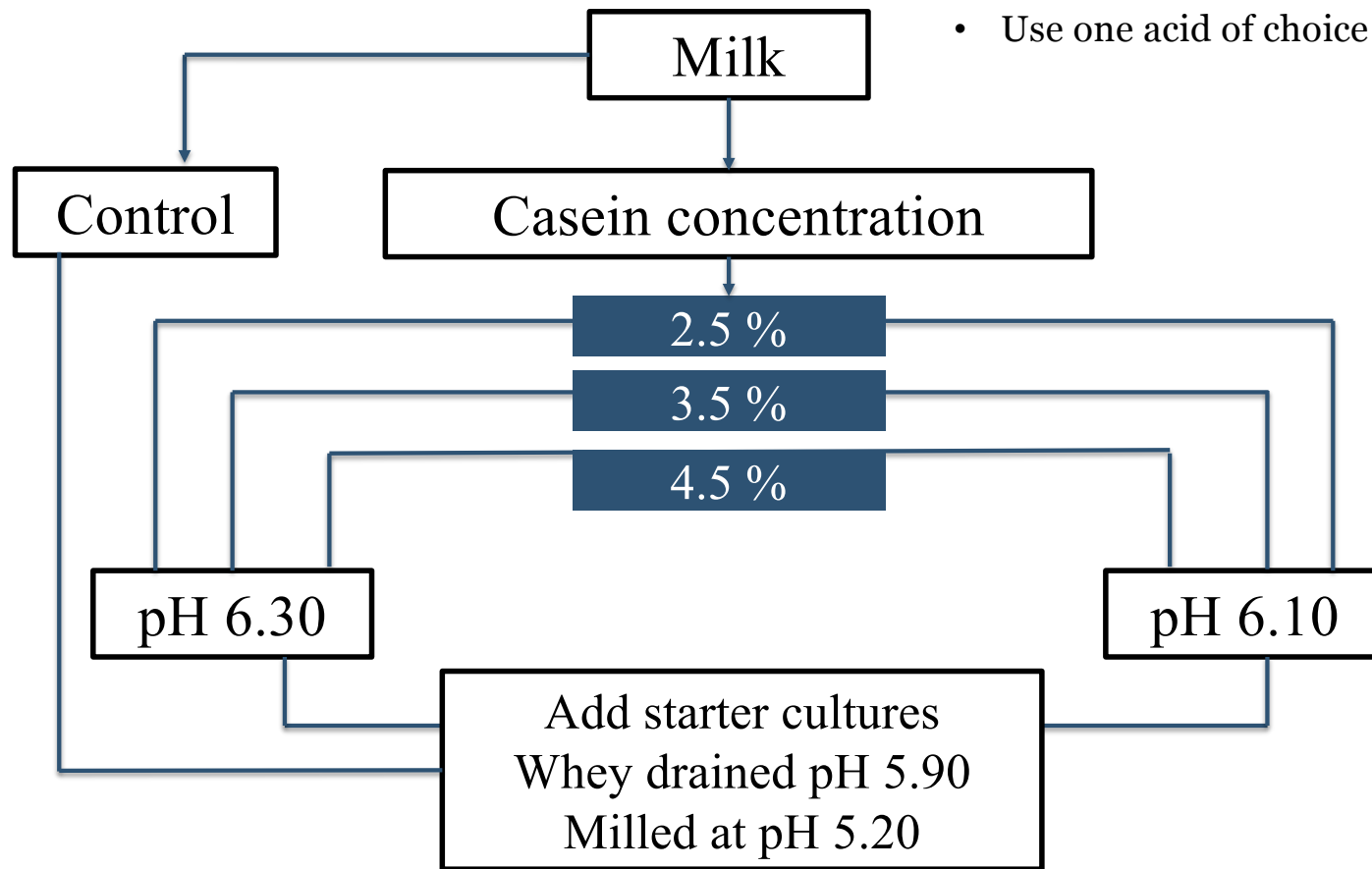
- Use glucono lactone for pre-acidification
- Control time using temperature
  - Higher degree of hydrolysis at high temp, faster acidification

## Exp 2: Effect of different acids vs pre-acidification pH



- If time is a variable from previous exp, the acidification time should be kept constant for all acids

# Exp 3: Effect of casein concentration vs pre-acidification pH



- Use one acid of choice for pre-acidification



# Analyses

- Calcium balances during cheese make
  - Milk samples (Initial milk, pre-acidified milk)
  - Curds (At whey drain, milled curd, cheese after brining)
- Analyses during storage (@ 2w and 1m)
  - Functionality tests (Rheology, Sensory)
  - pH, Calcium balances during storage (Also @ 1w)
  - Proteolysis
  - Confocal microscopy (Fiber structure)
  - Texture Profile Analysis (TPA)





# Ultimate Goals

- To develop predictive models for how rate of acidification during cheesemaking impacts insoluble calcium levels in cheese (both at end of manufacture and during ripening)
  - Relate these models to cheese functionality
  - Allow cheesemakers to select make conditions needed to achieve specific functionality (an app or spreadsheet)



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# *Thank You* to Our Supporters

Wisconsin and US Dairy Farm Families | Dairy Farmers of Wisconsin

National Dairy Council | CDR Industry Team | WCMA



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